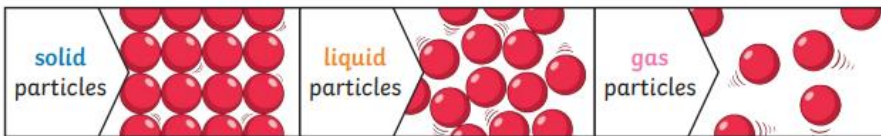




Vocabulary		
solids	noun	One of the three states of matter. Solid particles are very close together, meaning solids, such as wood and glass, hold their shape.
liquids	noun	This state of matter can flow and take the shape of the container because the particles are more loosely packed than solids and can move around each other. Examples of liquids include water and milk.
gases	noun	One of the three states of matter. Gas particles are further apart than solid or liquid particles and they are free to move around. A gas fills its container, taking both the shape and the volume of the container. Examples of gases are oxygen and helium.
evaporating	verb	When a liquid turns into a gas or vapour.
condensing	verb	When a gas, such as water vapour, cools and turns into a liquid.
conductor	noun	A material that heat or electricity can easily travel through. Most metals are both thermal conductors (they conduct heat) and electrical conductors (they conduct electricity).
insulator	noun	An insulator is a material that does not let heat or electricity travel through them. Wood and plastic are both thermal and electrical insulators.



Learning Question:
What would you need to be a CSI investigator?

Changes of State

solid 	The solid melts.		liquid
	The liquid freezes.		gas
liquid 	The gas condenses.		liquid
	The liquid evaporates.		gas

Scientific Enquiry Focus:
Observing changes over time

Key Knowledge

Reversible changes, such as mixing and dissolving **solids** and **liquids** together, can be reversed by:

<p>Sieving</p> <p>Smaller materials are able to fall through the holes in the sieve, separating them from larger particles.</p>	<p>Filtering</p> <p>The solid particles will get caught in the filter paper but the liquid will be able to get through.</p>	<p>Evaporating</p> <p>The liquid changes into a gas, leaving the solid particles behind.</p>
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Different materials are used for particular jobs based on their properties: electrical conductivity, flexibility, hardness, insulators, magnetism, solubility, thermal conductivity, transparency.

Dissolving
A solution is made when **solid** particles are mixed with **liquid** particles. **Materials** that will dissolve are known as soluble. **Materials** that won't dissolve are known as insoluble. A suspension is when the particles don't dissolve.

Sugar is a soluble **material**.

Sand is an insoluble **material**.



Irreversible changes often result in a new product being made from the old **materials** (reactants). For example, burning wood produces ash. Mixing vinegar and milk produces casein plastic.